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the following equilibrium given the results of an analysis of the equilibrium mixture which show that the concentration of  $\text{SO}_2$  is  $2.30 \times 10^{-3} \text{ mol dm}^{-3}$  and  $\text{SO}_3$  is  $3.92 \times 10^{-2} \text{ mol dm}^{-3}$ . Use your knowledge of ionic equations to give the molar ratio in which the following acids react with bases. Complete the table to show your answers. acid formula of acid base formula of base

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